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Acremoxanthones A and B, novel antibiotic polyketides from the fungus *Acremonium* sp. BCC 31806

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ABSTRACT

Acremoxanthones A and B, novel anthraquinone–xanthone heterodimers with a unique linkage pattern, together with two known compounds, acremonidins A and C, were isolated from the fungus *Acremonium* sp. BCC 31806. The structures of the acremoxanthones were determined by analysis of 2D NMR and mass spectrometric data. Acremoxanthones and acremonidins exhibited antibacterial, antifungal, antiplasmo-dial, and cytotoxic activities.

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As part of our research program on novel bioactive fungal secondary metabolites, we have investigated the constituents of the fungus *Acremonium* sp. BCC 31806 as the ¹H NMR spectrum of a mycelial extract from this fungus had shown unique resonance patterns.¹ Chemical studies on a large-scale fermentation broth of BCC 31806 afforded two new heterodimers of aromatic polyketides, acremoxanthones A (1) and B (2), along with two known compounds acremonidins A (3) and C (4).^{2,3} Details of the isolation, structure elucidation, and biological activities of these compounds are presented here.

Acremonium sp. BCC 31806 was fermented in potato dextrose broth (PDB; 20×250 mL) at 25 °C for 28 days under static conditions. The MeOH extract (1.31 g) from mycelia was subjected to fractionation using Sephadex LH-20 and silica gel column chromatography, and the eluted fractions containing products were further purified by preparative HPLC using a reverse phase column to furnish **1** (71 mg), **2** (26 mg), **3** (103 mg), and **4** (75 mg).

The molecular formula of acremoxanthone A (1) was determined to be $C_{33}H_{24}O_{11}$ by HRMS (ESI-TOF) and by analysis of ¹H, ¹³C, and DEPT NMR data.⁴ The IR spectrum (KBr) shows intense and broad absorption bands at v_{max} 1731 and 1641 cm⁻¹ and a broad hydroxy absorption at 3425 cm⁻¹. The ¹³C NMR spectrum in CDCl₃ shows 33 resolved peaks. DEPT and HMQC experiments enabled the categorization of these carbons into 19 sp² quaternary carbons, seven sp² methines, one sp³ quaternary carbon (δ_C 41.6),

one oxymethine ($\delta_{\rm C}$ 73.1), one sp³ methine ($\delta_{\rm C}$ 37.6), one methylene (δ_c 35.2), one methoxy group (δ_c 53.2), and two methyl groups ($\delta_{\rm C}$ 22.1 and 21.2). The ¹H NMR spectrum (CDCl₃) displayed four phenolic OH signals at $\delta_{\rm H}$ 14.15 (br), 12.81 (s), 11.62 (s), and 7.72 (br). Analysis of the HMBC data revealed the complete structure of **1** as the heterodimer of an anthraquinone moiety (ABC-ring) and a xanthone moiety (DEF-ring). Thus, the A-ring of the anthraquinone was established from the presence of two meta-coupled methines, H-5 and H-7, and the HMBC correlations from the chelated OH ($\delta_{\rm H}$ 11.62, 8-OH) to C-7, C-8, and C-8a, and from the $\delta_{\rm H}$ 2.40 methyl protons (br s, H-11) to C-5, C-6, and C-7. Weak 4-bond correlations from H-5 and H-7 to the $\delta_{\rm C}$ 186.1 quaternary carbon were observed, hence, this carbon (C-9) should be attached to the quaternary carbon C-8a. The C-ring was composed of a cis-olefin (C-3, C-4) connected to a methine (δ_C 37.6, C-2) and a quaternary carbon ($\delta_{\rm C}$ 41.6, C-4a). These carbons, together with two sp² carbons at $\delta_{\rm C}$ 185.9 (C-1) and $\delta_{\rm C}$ 105.2 (C-9a), formed a six-membered ring. This connectivity was shown by the COSY correlations (H-2-H-3-H-4) and HMBC correlations from H-2 to C-1 and C-9a, from H-3 to C-4a, and from H-4 to C-2, C-4a, and C-9a. The ABCring juncture was confirmed by the HMBC correlations from the oxymethine at $\delta_{\rm H}$ 6.02 (s, H-10) to C-4, C-4a, C-9a, C-10a, C-5, and C-8a. The xanthone structure (DEF-ring) was also deduced from the HMBC correlations. The D-ring portion was assigned from the HMBC correlations from a chelated OH ($\delta_{\rm H}$ 12.81, 1'-OH) to C-1', C-2', and C-9'a, from H-4' to C-2', C-4'a, and C-9'a, and from the non-equivalent methylene protons ($\delta_{\rm H}$ 2.93 and 2.80, H-11') to C-2', C-3', and C-4'. A weak 4-bond HMBC correlation from H-4' to

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the $\delta_{\rm C}$ 180.2 quaternary carbon and chelated 1'-OH ($\delta_{\rm H}$ 12.81) indicated the carbonyl linkage at C-9'. The rest of the ¹H and ¹³C NMR signals were assigned to those for the F-ring portion, which consisted of a tetrasubstituted benzene with vicinally coupled methines at $\delta_{\rm H}$ 7.38 (br d, J = 9.1 Hz, H-5') and 7.34 (br d, J = 9.1 Hz, H-6') and substituted with one OH and one methoxycarbonyl group. The molecular formula of 1, established by HRMS, and the upfield ketone carbonyl (C-9', $\delta_{\rm C}$ 180.2) resonance required the tricyclic xanthone structure (DEF-ring). The linkage between the two tricyclic units was addressed also from the HMBC correlations: from H-2 to C-1, C-2, and C-3, from H-10 to C-11', and from H-11' to C-4, C-4a, C-9a, and C-10. Proton H-4 and one of the diastereotopic methylene protons at δ_H 2.93 (H_β-11') show a weak NOESY cross-peak, whilst H-10 exhibited NOESY correlations to both H_{β} -11' and $H_{\alpha}\text{--}11'$ $(\delta_{H}$ 2.80) as well as weak cross-peaks with H-4 and H-5. These data indicated that H-10 and C-11' should be svnfacial.

The molecular formula of acremoxanthone B (**2**)⁵ was determined by HRMS as $C_{33}H_{24}O_{12}$, containing one more oxygen atom than **1**. The ¹H and ¹³C NMR data for **2** were similar, in particular the xanthone moiety, to those of **1**. The major differences were the down-field shifted C-1 (δ_C 205.2) and C-9 (δ_C 195.8) when compared to **1** (δ_C 185.9 and 186.1), and the replacement of the sp² quaternary carbon at δ_C 105.2 (C-9a) of **1** by a tertiary alcohol (δ_C

82.2). The chelated phenolic proton, 9-OH ($\delta_{\rm H}$ 14.15, br s) for 1, was absent in **2**. Instead, an OH resonated at $\delta_{\rm H}$ 7.69 (9a-OH), which exhibited HMBC correlations to C-4a and C-9a ($\delta_{\rm C}$ 82.2). Detailed analysis of the 2D NMR data, in particular HMBC correlations (Table 1), revealed that the xanthone part of the molecule was identical to 1. Therefore, acremoxanthone B (2) is the C-9a oxygenated derivative of 1. The relative configuration of 2 was addressed by analysis of NOESY data (DMSO- d_6). Proton H-4' showed NOESY correlations to H_{α} -11' (δ_{H} 3.34) and H_{β} -11' (δ_{H} 3.19) with slightly higher cross-peak intensity for the former, while H-10 showed a correlation to H_{α} -11' with a higher intensity than that to H_{β} -11'. Proton H-10 did not show clear NOESY correlations to H-5 or H-4. These data strongly suggested a pseudoaxial orientation of H-10. Unfortunately, there was no NOESY cross-peak from 9a-OH (br s) to any CH or CH₂ group. The NOESY spectrum acquired in acetone- d_6 also provided similar information. However, examination of a molecular model of both the $9a-\alpha$ -OH and $9a-\beta$ -OH isomers demonstrated that only isomer $9a-\alpha$ -OH is consistent with the above-mentioned NOESY data. In the $9a-\alpha$ -OH isomer, 9a-OH and H-10 occupy pseudoaxial positions (Fig. 1). In contrast, the 9a-β-OH configuration required pseudoequatorial orientation of H-10.

The structures of compounds **3** and **4** were elucidated on the basis of HRMS and 2D NMR data.^{6,7} The ¹H and ¹³C NMR spectral data for these isolated compounds in DMSO- d_6 were identical to those of acremonidins A and C, respectively.² Acremonidins A–E were previously isolated from *Acremonium* sp. *LL*-Cyan 416. In the original report,² the relative configuration of C-2, C-4a, and C-10 for acremonidins A and C was proposed as shown in the structures **3** and **4**, respectively, whereas the configuration of the additional chiral carbon center, C-9a, in **4** was not presented. Since the ¹H and ¹³C NMR data for the ABC-ring moiety are very close between acremoxanthone B (**2**) and acremonidin C (**4**) (in DMSO- d_6), and considering also their co-occurrence in BCC 31806, we propose that these compounds possess the same stereochemistry at C-9a.

A unique structural feature of acremoxanthones is the linkage pattern of two tricyclic units. Ring junctions similar to those of acremoxanthones/acremonidins are shown only in xanthoquinodins (anticoccidial antibiotics from *Humicola* sp. FO-888),⁸ beticolins/cebetins (yellow toxins produced by *Cercospora beticola*),^{9,10} and chaetomanone (from *Chaetomium globosum* KMITL-N0802).¹¹ Similar to the biosynthesis of xanthoquinodins,^{8b} acremoxanthones are probably formed via coupling between the xanthone (elminthosporin) and the xanthone (pinselin) moieties.¹² Xanthone and benzophenone monomers, such as those corresponding, respectively, to acremoxanthones and acremonidins, are known to be produced by oxidative cleavage of common anthraquinones.^{8b,12,13}

Compounds 1-4 were subjected to in vitro biological assays: antibacterial (Staphylococcus aureus, Bacilus cereus, Pseudomonas aeruginosa, and Escherichia coli),14 antifungal (Candida albicans, Magnaporthe grisea),¹⁴ antimycobacterial (Mycobacterium tuberculosis H37Ra), and antiplasmodial (Plasmodium falciparum K1)¹⁵ activities, and cytotoxicity to three cancer cell-lines (KB, BC, and NCI-H187)¹⁴ and non-cancerous Vero cells. Table 2 shows selected data. Compounds 1-4 exhibited moderate antibacterial activity against S. aureus and B. cereus, but they were inactive against P. aeruginosa and E. coli. Only acremoxanthone A (1) displayed significant activity against C. albicans (IC₅₀ 1.7 μ g/mL), while all the compounds were inactive against the plant pathogenic fungus M. grisea. All these compounds lacked antituberculosis activity; however, the antimalarial activity of 2 and 4 should be noted. Whereas compounds 1-4 exhibited significant antibiotic activities, they also showed cytotoxicity to all tested cell-lines. Acremonidins A (3) and B (deacetyl analogue of **3**) are known to show moderate activity against Gram-positive bacteria, including the methicillin-resistant staphylococci and vancomycin-resistant enterococci.²

able 1	
IMR data for acremoxanthones A (1) and B (2) (500 MHz for 1 H, and 125 MHz for 13 C))

Position	1 (in CDCl ₃)			2 (in DMSO- <i>d</i> ₆)			
	$\delta_{\rm C}$ mult.	$\delta_{\rm H}$ mult. (J in Hz)	НМВС	$\delta_{\rm C}$ mult.	$\delta_{\rm H}$ mult. (J in Hz)	HMBC	
1	185.9 qC			205.2 qC			
2	37.6 CH	4.94 dd (6.6, 0.6)	1,3,4,9a,1',2',3'	44.0 CH	4.69 d (6.1)	1,3,4,9a,1',2',3'	
3	131.9 CH	6.49 dd (8.4, 6.7)	2,4,4a,5	134.9 CH	6.54 dd (9.1, 6.6)	1,2,4a	
4	132.5 CH	6.13 dd (8.4, 0.6)	2,3,4a,9a,10	129.6 CH	5.84 d (9.1)	2,4a,9a,10,	
4a	41.6 gC			47.2 qC			
5	123.2 CH	6.93 br s	6,7,8a,9,10,10a,11	118.8 CH	6.65 br s	7,8a,10,11	
6	147.8 gC			150.0 gC			
7	119.4 CH	6.82 br s	5.8.8a.9.11	117.9 CH	6.84 brs	5.8a.11	
8	161.8 aC			163.2 gC		.,,	
8-0H	1	11.62 s	7.8.8a		11.40 s	7.8.8a	
8a	112.7 aC			110.4 aC			
9	186.1 gC			195.8 gC			
- 9-0H		14.15 br s					
9a	105.2 aC			82.2 aC			
9a-0H	room qe			on de	7 69 s	4a 9a	
10	73 1 CH	6.02 s	4 4a 5 8a 10a 12 11′	70 5 CH	6 33 br s	4 4a 10a 12 11'	
10a	136.5 aC	0102 0	1, 14,0,04,104,12,11	140.0 aC	0.00 01 0	1, 14,104,12,11	
11	22.1 CH ₂	2 40 s	567	22.4 CH ₂	2 37 br s	567	
12	170.6 aC	2110 0	51017	171.5 aC		5, 6, 7	
13	21.2 CH ₂	2 03 s	12	21 4 CH ₂	2 38 s	12	
1/	157.8 aC	2.03 3	12	156.7 aC	2.50 5	12	
1′-ОН	157.0 qe	12.81 s	1' 2' 9' 2	150.7 qe	12.83 s	1' 2' 9' 3	
2'	117.4 aC	12.01 3	1,2,3 u	118.4 aC	12.05 5	1,2,5 u	
3/	146.4 gC			147.1 aC			
<u>4</u> ′	109.8 CH	6 47 br s	2' 4'a 9' 9'a 11'	1099 CH	6 99 s	2' 4'a 9'a 11'	
4'a	154.0 aC	0.17 51 5	2,10,5,50,11	154.3 aC	0.55 5	2,10,00,11	
5/	122.0 CH	7.38 br d(9.1)	7/ 8/3 10/3	120.6 CH	7.61 d (9.2)	7/8/2	
5 6'	125.7 CH	7.36 br d (9.1)	8/ 10/a	126.0 CH	7.61 d (9.2)	8/ 10/ a	
0 7'	152 3 ^a aC	7.54 bi d (5.1)	0,104	151.3° aC	7.40 u (5.2)	0,10 a	
, 7′-0H	152.5 qc	7 72 br s		151.5 qc	10.53 br s		
8'	114.2 aC	7.72 01 3		117 50 ^b aC	10.55 01 5		
8/a	118.4 aC			117.50 qC			
0 a 0/	180.2 dC			180.0 aC			
9 9/1	100.2 qC			106.2 qC			
5 a 10/ 2	150.6ª aC			$140.4^{\circ} aC$			
10 d 11/	25.2 CU	202 d(177)	4 42 02 10 2/ 4/	145.4 qC	224 d (102)	1 1 2 1 1	
11	55.2 CH ₂	2.95 u(17.7)	4,44,94,10,2,4	52.5 CH2	2.34 u (19.2)	4,4d,5,4	
10/	160.0 cC	2.00 u (17.7)	4,4d,9d,10,2 ,3 ,4	167.1 cC	3.19 u (19.2)	4d,9d,2 ,5	
12	109.0 qC	4.01 c	12/	107.1 qC	2.95 a	12/	
15	ЭЭ.2 CП3	4.01 5	12	52.7 CH ₃	2.02.5	12	

^{a-c} Assignment of carbons can be interchanged.



Figure 1. Probable stereo structure of acremoxanthone B (2).

Table	2

Biological activities of compounds 1-4

Compound	S. aureus ^a	B. cereus ^a	C. albicans ^b	P. falciparum ^c		Cytotoxicity (IC ₅₀ , µg/mL) ^d			
	(MIC, µg/mL)	(MIC, µg/mL)	(IC ₅₀ , μg/mL)	(IC ₅₀ , μg/mL)	KB	BC	NCI-H187	Vero	
1	12.5	>100	1.7	>10	3.3	1.1	0.87	1.2	
2	6.25	6.25	>50	3.0	18	9.4	14	12	
3	3.13	1.56	>50	5.4	13	5.0	4.5	6.6	
4	3.13	3.13	>50	>10	>20	9.3	16	23	

^a Antibacterial activity against Staphylococcus aureus and Bacillus cereus was evaluated using the resazurin microplate assay (REMA). MIC values of a standard antibacterial ^b Antifungal activity against *Candida albicans*. Standard compound, amphotericin B, showed an IC₅₀ value of 0.047 µg/mL.
^c Antimalarial activity against *Plasmodium falciparum* K1. Standard antimalarial drug, dihydroartemisinin, showed an IC₅₀ value of 0.0011 µg/mL.
^d The IC₅₀ values of a standard compound, ellipticine, against KB, BC, NCI-H187 and Vero cells were 0.27, 0.21, 0.15 and 0.60 µg/mL, respectively.

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- 1. This fungus was isolated from indoor air at BIOTEC, Thailand, and is deposited at the BIOTEC Culture Collection as BCC 31806.
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- 4. Acremoxanthone A (1): Yellow powder; $[\alpha]_{26}^{26}$ +649 (c 0.20, CHCl₃); UV (MeOH) λ_{max} (log ε) 211 (4.26), 245 (4.46), 273 (4.47), 330 (4.18), 366 (4.37) nm; IR (KBr) v_{max} 3425, 1731, 1641, 1612, 1577, 1220, 1015 cm⁻¹; HRMS (ESI-TOF) m/z 595.1277 [M–H]⁻ (calcd for C₃₃H₂₃O₁₁, 595.1246); ¹H and ¹³C NMR data in CDCl₃, Table 1.
- 5. Acremoxanthone B (2): Yellow powder; $[\alpha]_D^{27}$ +52 (*c* 0.10, CHCl₃); UV (MeOH) λ_{max} (log ε) 216 (4.12), 241 (4.21), 275 (4.45), 302 (sh) (4.00), 397 (sh) (3.59) nm; IR (KBr) ν_{max} 3443, 1737, 1639, 1433, 1233, 1211 cm⁻¹; HRMS (ESI-TOF) m/z 635.1175 [M+Na]⁺ (calcd for C₃₃H₂₄O₁₂Na, 635.1160); ¹H and ¹³C NMR data in DMSO-d₆), Table 1.
- 6. Acremonitian A (3): Yellow powder; $[\alpha]_D^{26}$ +422 (*c* 0.20, CHCl₃); UV (MeOH) λ_{max} (log ε) 208 (4.04), 241 (4.23), 276 (4.15), 290 (sh) (4.11), 362 (4.29) nm; IR (KBr) ν_{max} 3426, 3204, 1731, 1681, 1614, 1462, 1217 cm⁻¹; HRMS (ESI-TOF) *m/z* 637.1311 [M+Na][†] (calcd for C₃₃H₂₆O₁₂Na, 637.1316); ¹H and ¹³C NMR spectral data in DMSO-*d*₆ were identical to those reported in the literature.
- data in DMSO- d_6 were identical to those reported in the literature. 7. Acremonidin C (**4**): Yellow powder; $[\alpha]_{2}^{26} - 20 (c 0.10, CHCl_3); UV (MeOH) \lambda_{max}$ (log ε) 212 (4.17), 241 (4.33), 278 (4.43), 349 (4.18) nm; IR (KBr) v_{max} 3426,

3208, 1736, 1680, 1624, 1207 cm⁻¹; HRMS (ESI-TOF) m/z 629.1318 [M–H]⁻ (calcd for $C_{33}H_{25}O_{13}$, 629.1301); ¹H and ¹³C NMR spectral data in DMSO- d_6 were identical to those reported in the literature.

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